

**REFRACTORY ARTICLES****Publication number:** EP1282477 (A1)**Publication date:** 2003-02-12**Inventor(s):** BELL DAVID ALAN [GB]; JONES DAVID LLEWELLYN [GB]; JUMA KASSIM [GB]; ANSORGE ANDREAS [DE]; SCHMIDT MARTIN [DE]**Applicant(s):** FOSECO INT [GB]**Classification:****- international:** *B22C9/08; B22D43/00; C04B38/00; B22C9/00; B22D43/00; C04B38/00*; (IPC1-7): B22C9/08; B22D43/00; C04B38/00; C04B38/06**- European:** B22C9/08C; B22D43/00R2; C04B38/00A; C04B38/00H2**Application number:** EP20010960970 20010828**Priority number(s):** WO2001GB03846 20010828; GB20000021343 20000831; GB20000022676 20000915; GB20000025411 20001017; GB20010007872 20010329**Also published as:**

EP1282477 (B1)



EP1282477 (B8)



EP1282477 (B2)



WO0218075 (A1)



US2004128857 (A1)

more &gt;&gt;

Abstract not available for EP 1282477 (A1)

Abstract of corresponding document: **WO 0218075 (A1)**

A filter for molten metal is an open-pored porous material comprising particles of refractory material embedded in and bonded together by a carbon matrix bonding material. The filter can be made by forming a porous article from refractory particles, e.g. refractory oxide, carbide or graphite, and a carbon-rich binder, e.g. tar, pitch or an organic (preferably aromatic) polymer that degrades to form carbon on pyrolysis, and then firing the porous article to generate the carbon matrix in which the refractory particles are embedded. The porous article is preferably made by coating a reticulated polyurethane foam with binder and refractory particles, and firing at preferably no higher than 800 DEG C.

---

Data supplied from the **esp@cenet** database — Worldwide

(19) World Intellectual Property Organization  
International Bureau



(43) International Publication Date  
7 March 2002 (07.03.2002)

PCT

(10) International Publication Number  
**WO 02/18075 A1**

(51) International Patent Classification<sup>7</sup>: **B22C 9/08**,  
B22D 43/00, C04B 38/00, 38/06

MX, MZ, NO, NZ, PH, PL, PT, RO, RU, SD, SE, SG, SI,  
SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU,  
ZA, ZW.

(21) International Application Number: PCT/GB01/03846

(22) International Filing Date: 28 August 2001 (28.08.2001)

(25) Filing Language: English

(26) Publication Language: English

(84) Designated States (*regional*): ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).

(30) Priority Data:

0021343.9	31 August 2000 (31.08.2000)	GB
0022676.1	15 September 2000 (15.09.2000)	GB
0025411.0	17 October 2000 (17.10.2000)	GB
0107872.4	29 March 2001 (29.03.2001)	GB

**Declarations under Rule 4.17:**

- as to applicant's entitlement to apply for and be granted a patent (Rule 4.17(ii)) for the following designations AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PH, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, UZ, VN, YU, ZA, ZW, ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG)
- as to the applicant's entitlement to claim the priority of the earlier application (Rule 4.17(iii)) for all designations
- of inventorship (Rule 4.17(iv)) for US only
- of inventorship (Rule 4.17(iv)) for US only

(71) Applicant (*for all designated States except US*): **FOS-ECO INTERNATIONAL LIMITED** [GB/GB]; Burnah Castrol House, Pipers Way, Swindon, Wiltshire SN3 1RE (GB).

(72) Inventors; and

(75) Inventors/Applicants (*for US only*): **BELL, David, Alan** [GB/GB]; 2 The Glen, Stone, Stafford ST15 0HB (GB). **JONES, David, Llewellyn** [GB/GB]; 97 Bittell Road, Barnt Green, Birmingham B45 8LX (GB). **JUMA, Kassim** [GB/GB]; 11 Exeter Street, Stafford ST17 4BG (GB).

(74) Agent: **MOSEY, Stephen, George**; Marks & Clerk, Alpha Tower, Suffolk Street, Queensway, Birmingham B1 1TT (GB).

**Published:**

- with international search report
- before the expiration of the time limit for amending the claims and to be republished in the event of receipt of amendments

(81) Designated States (*national*): AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW,

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

(54) Title: REFRACTORY ARTICLES

(57) Abstract: A filter for molten metal is an open-pored porous material comprising particles of refractory material embedded in and bonded together by a carbon matrix bonding material. The filter can be made by forming a porous article from refractory particles, e.g. refractory oxide, carbide or graphite, and a carbon-rich binder, e.g. tar, pitch or an organic (preferably aromatic) polymer that degrades to form carbon on pyrolysis, and then firing the porous article to generate the carbon matrix in which the refractory particles are embedded. The porous article is preferably made by coating a reticulated polyurethane foam with binder and refractory particles, and firing at preferably no higher than 800 °C.

WO 02/18075 A1

### REFRACTORY ARTICLES

The present invention relates to refractory articles and to a process for producing refractory articles. The invention particularly relates to refractory articles which are suitable for use in contact with molten metals (including metal alloys) at elevated temperatures. In particular, the present invention relates to a filter suitable for filtering  
5 molten metal and to a process for producing such a filter.

Refractory articles (i.e. articles formed from refractory materials) such as vessels, vessel linings, sleeves, pouring cups, filters, moulds, mould cores, etc are used in the handling, pouring and casting of molten metals. Such refractory articles commonly need to be fired at very high temperatures, for example in the region of  
10 1000°C or higher. Firing at such high temperatures is generally costly and time-consuming. It would therefore be highly desirable to reduce the required firing temperatures of such articles.

Refractory materials comprising carbon, e.g. in the form of complex structures bonded with carbon, have many useful applications, but the manufacture of such  
15 materials by known techniques has many associated problems and limitations

Known techniques for making carbon-bonded articles include the graphitisation of petroleum pitch and coal tar. These processes require conditions of very high pressure followed by heating to high temperatures in the order of 2,500°C. Such a process is energy inefficient, provides hazardous conditions and can only satisfactorily  
20 produce material in sections of several inches in size, e.g. in cross section.

An alternative technique is the direct hot mixing of pitch and tar with other mineral powders such as graphite, refractory powders etc. followed by pressing at high pressure prior to tempering for several days in specially designed kilns at temperatures

in the order of 1,400°C. Again such a technique is energy inefficient and can only produce materials of thick section.

5 A further known technique is the high energy mixing of synthetic resin with mineral powders or graphite in the presence of a cross-linker which results in the conversion of the resin from a thermoplastic to a thermosetting polymer. Subsequent heating in the absence of air to a temperature up to 1,000°C yields a material which again can only be manufactured with thick cross-section.

10 A still further process is the use of a sinterable carbon precursor containing meso-phase which on heating to a temperature of up to 1,000°C can yield a carbon bond structure. However the sinterable carbon used with the precursor is very expensive to manufacture.

EP-A-0708064 discloses the manufacture of fire resistant carbon compound blocks with a magnesium oxide base which are used, for example, for lining furnaces and the vessels used to receive smelted steel. The source of carbon is suitably  
15 mesogenous pitch, or polyaromatic compounds mesophase, or mesophase pitch.

DE-A-4307343 discloses binding and impregnating pitches for the manufacture of carbon electrodes and fire resistant materials. The pitches are produced by heating oils, tars or pitches derived from coal or crude oil at a temperature in the range 360 to 540°C in a substantially inert gas atmosphere at elevated pressure.

20 WO 99/28273 (Foseco International Limited) discloses a filter for molten metal comprising a porous carbon foam substrate coated substantially throughout with a refractory metal or refractory compound formed by chemical vapour deposition. The filter can be made by pyrolysing a porous organic foam substrate, for example, polyurethane foam or resin-coated polyurethane foam, at a temperature in the range 600  
25 to 1200°C. The porous carbon foam is then coated by vapour phase deposition with a refractory material (e.g. as described in US Patents Nos. 5154970, 5283109 and 5372380 and in EP-A-0747124.

On the other hand, US Patent No 5104540 (Corning Incorporated) discloses a carbon-coated porous sintered ceramic filter for molten metal comprising a monolith  
30 substrate formed from inorganic refractories such as alumina, mullite, zircon, zirconia, spinel, cordierite, lithium aluminosilicates, titania, feldspars, quartz, fused silica, silicon carbide, kaolin clay, aluminium titanate, silicates, aluminates and mixtures thereof. The carbon coating is applied to the surface of the pre-formed filter or to a thermite coating

thereon.

A type of filter for filtering molten metal, which has been extensively employed in the art, comprises an open-cell porous structure comprising refractory particles bound together by an inorganic matrix. Filters of this type can be made, for example, by coating a suitable open-celled foam, e.g., reticulated polyurethane foam, with a composition comprising a slurry of refractory particles (e.g. zirconia), a binder (e.g. a borosilicate glass) and water, drying the coated foam and then firing to produce an open-celled porous filter comprising the refractory particles bound together by a glass matrix. The polyurethane foam generally chars and burns away during the firing process and does not contribute significantly to the properties of the produced filter. Filters of this type are disclosed, for example, in EP-A-0412673 and EP-A-0649334 both in the name of Foseco International Ltd.

It is an object of the present invention to provide an improved filter for filtering molten metal. It is another object of the present invention to provide a process for preparing a filter for filtering molten metal.

Accordingly the present invention comprises a filter suitable for filtering molten metal comprising an open-pored porous material comprising particles of refractory material embedded in and bonded together by a bonding material comprising a carbon matrix.

The phrase "open-pored porous material" means throughout this specification a material comprising a regular, partly regular, irregular or random distribution of pores within a solid material, the pores providing channels through which a molten metal may flow through the said material. The pores can be fully or partially inter-communicating or can be provided by a plurality of passages which traverse the bulk of the said material to enable molten metal to flow readily there-through. The pores themselves can be of regular or irregular size and shape. For example, the pores can be provided by a series of parallel ducts passing linearly through the material, the ducts having any desired cross-section, for example, circular, elliptical, triangular, square or pentagonal. Or, for example, the pores can be provided by a random distribution of irregular interconnecting passages, for example, similar to the pore distribution in a natural sponge. A preferred open-pore porous material is one having a fairly regular pore distribution provided by commercially available reticulated open cell polyurethane foams. Such materials are well known in the art for use in the manufacture of refractory

filters for the filtration of molten metals.

The refractory material bonded by the bonding material in the present invention can be, for example, any of the well known refractory materials in the art which provide resistance to the corrosive effect and high temperature of the molten metal which it is  
5 desired to filter. Examples of refractory materials suitably used in the present invention include: zirconia, zircon, silica, alumina, titania, carbides (e.g. silicon carbide, zirconium carbide, titanium carbide, calcium carbide, aluminium carbide), nitrides (e.g. silicon nitride and aluminium nitride), metal oxides (e.g. nickel oxide and chromic oxide), magnesia, mullite, graphite, anthracite, coke, active carbon, graphite-refractory  
10 (e.g. graphite-magnesia, graphite-alumina, graphite-zirconia), or mixtures comprising two or more of these.

The bonding material comprises a carbon matrix which bonds the refractory particles together and in which the particles are embedded in the filter of the present invention. The bonding material is preferably in the form of a coke or semicoke which  
15 is produced by the thermal decomposition of an organic material.

The relative proportions (in percentage by weight) of particulate refractory material to bonding material are preferably in the range at least 50% refractory : no more than 50% bonding material; more preferably they are in the range at least 55% refractory : no more than 45% bonding material; even more preferably they are in the  
20 range at least 60% refractory : no more than 40% bonding material, for example approximately 65-75% refractory : approximately 35-25% bonding material.

Filters in accordance with the present invention suitably used for filtering molten steel preferably comprise refractory particles selected from zirconia, zircon, silicon carbide, graphite, alumina and mixtures or two or more thereof.

25 Filters in accordance with the present invention suitably used for filtering molten iron preferably comprise refractory particles selected from zirconia, zircon, silicon carbide, graphite, alumina, alumino-silicates (e.g. chamotte, pyrophyllite, andalusite) and mixtures of two or more thereof

Filters in accordance with the present invention suitably used for filtering  
30 molten aluminium preferably comprise refractory particles selected from graphite and wollastonite.

The present invention further comprises a process for making a material suitable for filtering molten metal comprising forming into an open-pored porous

material a mixture comprising a binder and refractory particles and firing the material, wherein the binder is a carbon-rich source selected from one or more of the following: pitch, tar, and aromatic organic polymer that degrades to form carbon on pyrolysis.

Thus for example, an open-pored porous material can be formed by  
5 compressing a mixture of the binder and the refractory particles into a disk or slab in a die and piercing the compressed material with a plurality of needles or rods to create perforations traversing the thickness of the disc or slab. The perforations may have any desired cross-section and are preferably, for example, circular, elliptical, triangular, square or pentagonal. Preferably the piercings are arranged in a regular grid pattern  
10 across the surface of the article. A similar type of article can be produced for example by extruding a mixture comprising the binder and the refractory particles, preferably together with a fluid and/or other additives to render the mixture more readily extrudable, through a suitable die equipped with a plurality of mandrels to form perforations in the extruded article. Extrusion of articles having this type of shape is  
15 well known in the art.

A preferred method in accordance with the present invention for making an open-pored porous material suitable for filtering molten metal comprises

- (1.) forming a slurry comprising (a) particles of a refractory material, (b) a binder and (c) a liquid carrier,
- 20 (2.) coating a disposable former with the slurry
- (3.) drying the coated former,
- (4.) optionally applying one or more additional coats of a refractory material and/or a binder, optionally with liquid carrier, and drying the one or more additional coats,
- 25 (5.) firing the coated former to produce a porous material,

wherein the binder is a carbon-rich source selected from one or more of the following classes of materials: pitches, tars, and organic polymers that degrade to form carbon on pyrolysis.

A further preferred method in accordance with the present invention of making  
30 an open-pored porous material suitable for filtering molten metal comprises

- (1.) forming a slurry comprising (a) particles of a refractory material, (b) a binder and (c) a liquid carrier,
- (2.) coating a disposable former with the slurry

- (3.) drying the coated former,
  - (4.) optionally applying one or more additional coats of a refractory material and/or a binder, optionally with liquid carrier, and drying the one or more additional coats,
  - 5 (5.) firing the coated former to produce a porous material,
- wherein the binder is a carbon-rich source selected from one or more of the following classes of materials: pitches, tars, and organic polymers that degrade to form carbon on pyrolysis, and wherein (i) the binder has been subjected to stabilisation by pre-treatment with an acid and/or an oxidising agent and/or (ii) a polyfunctional compound
- 10 is included in the slurry or the coating to promote stabilisation of the binder.

The carbon-rich source is selected from one of more classes of material which comprise pitches, tars and polymers that degrade to form carbon on pyrolysis. Examples of such materials are coal tar, petroleum pitch, asphalt, bitumen, synthetic pitch, synthetic tar, synthetic bitumen; or residues derived from the pyrolysis of coal, crude

15 oil, coal tar, petroleum pitch, asphalt, bitumen, synthetic pitch, synthetic tar or synthetic bitumen. The polymers which are degradable to form carbon on pyrolysis are preferably aromatic organic polymers. Aromatic organic polymer can suitably be any polymer, synthetic or natural, that contains organic aromatic chains, aromatic networks or aromatic substituents, and that degrades on pyrolysis to form a carbonaceous residue.

20 The aromatic chains or substituents are preferably based on benzene rings which may be, for example, pendant from a molecular chain or may be present as part of a molecular chain or network. Examples of suitable aromatic polymers are phenyl substituted polymers, ortho-, meta- or para-phenylene polymers, naphthalene polymers, phenanthrene polymers, anthracene polymers, coronene polymers and polymers of like

25 high molecular weight polynuclear aromatic materials. Polymers may be derived from furfuryl alcohol and these can also be employed. Preferred aromatic polymers are phenolic resins derived from the reaction of phenol and formaldehyde. Examples of such resins are resole resins and novolak resins which are well known in the art. Preferably the aromatic polymers are crosslinkable or crosslinked polymers. The

30 aromatic polymers are thermosetting resins in preference to thermoplastic resins.

The refractory materials suitably used in the present invention are already recited above and include zirconia, zircon, silica, alumina, titania, carbides (e.g. silicon carbide, zirconium carbide, titanium carbide, calcium carbide, aluminium carbide),

nitrides (e.g. silicon nitride and aluminium nitride), metal oxides (e.g. nickel oxide and chromic oxide), magnesia, mullite, graphite, anthracite, coke, active carbon, graphite-refractory (e.g. graphite-magnesia, graphite-alumina, graphite-zirconia), or mixtures comprising two or more of these.

5           The particulate refractory material can, for example, comprise a mixture of graphite and another refractory material (e.g. alumina). For example the graphite content of the refractory particles is preferably from zero to 50 weight %, more preferably from 10 to 40 weight % based on the total weight of the refractory particles, the balance being met by one or more other particulate refractory materials.

10           The particles of refractory material employed in the process of the present invention can be, for example, powders, fines, granules, fibrous materials, or microspheres (hollow and/or solid). Substantially any refractory material filler may be used; the skilled person will be able to select the appropriate material or mixture of materials according to the particular use requirements of the refractory composition or  
15   article.

          The particle size and the particle size distribution of the refractory particles can vary widely. Generally it is preferred that the particles have a mean particle size less than 30 microns, more preferably less than 10 microns, even more preferably in the range 1-5 microns.

20           The relative proportions (in percentage by weight) of the particulate refractory material to binder are preferably in the range at least 50% refractory : no more than 50% binder; more preferably they are in the range at least 55% refractory : no more than 45% binder; even more preferably they are in the range at least 60% refractory : no more than 40% binder, for example approximately 65-75% refractory : approximately 35-25%  
25   binder

          The binder used in the process of the present invention is preferably a carbon-rich material that provides a good yield of coke when heated to temperatures in the range of 500 to 700°C.

30           Preferably the binder is a particulate material. The particle size and the particle size distribution of the binder can vary widely. Generally it is preferred that the particles have a mean particle size less than 50 microns, more preferably less than 30 microns.

          The thermal decomposition of coal tar and pitch has been well documented [see for example Rand, B.McEnaney: "Carbon Binders from Polymeric Resins and Pitch

Part I – Pyrolysis Behaviour and Structure of the Carbons” Br. Ceram. Trans. J.,84, 57-165 (1985) no. 5]. During firing the pitch first melts then starts to pyrolyse, low molecular weight species are evolved and condensation reactions occur where the aromatic rings of less volatile constituents start to join up (polymerise). On further heating between 300 and 400°C large sheet-like structures of these polyaromatics are formed which then start stack up on top of one another. The sheets in the stacks are held together by Van der Waals forces. These stacked systems form liquid crystals and separate out of the rest of the liquid pitch as disks or spheres. This liquid crystal phase is called ‘mesophase’ or intermediate phase. Both coal tar and petroleum pitches produce mesophase, the amount produced being primarily dependent upon the aromatic content of the starting material. As heating progresses more and more mesophase is produced and the viscosity of the system starts to rise until around 500 – 550°C a solid mesophase or ‘semicoke’ is formed. This is the bonding phase, i.e. carbon bond in the refractory system. Factors which promote the formation of the semicoke, i.e. increasing the coke yield include; a high aromatic content of the pitch (coal tar has a greater aromatic content than petroleum pitch), reducing the amount of volatiles lost during the initial heating and promoting the reactions which join the aromatic molecules together. The latter can be achieved by either oxidising the pitch, which promotes dehydrogenation and the joining together (polymerisation) of aromatic rings, or reacting it with chlorine or sulphur which combine with hydrogen, thus removing the hydrogen from the system as HCl or H<sub>2</sub>S.

In the present invention the binder is preferably produced from a starting material comprising one or more organic compounds, for example, crude oil, coal or a synthetic aromatic polymer (e.g. phenolic resin). The binder preferably comprises a treated or untreated pitch and/or tar (e.g. coal tar) and/or synthetic phenolic resin (e.g. novolac or resole). The pitch and/or tar and/or synthetic phenolic resin is preferably in a powder or granular form. The pitch and/or tar and/or synthetic phenolic resin may be treated in order to create the mesophase in the pitch and/or tar and/or synthetic phenolic resin, or to increase the proportion of mesophase already present in the pitch and/or tar and/or synthetic phenolic resin.

The starting binder material incorporated in admixture with the refractory particles or in the slurry used to make the filter of the present invention can, if desired, be free from mesophase or can contain mesophase. Preferably an amount of mesophase

in the range 0-50 wt% based on total binder is present in the starting binder. During the firing of the filter, it is preferred that the binder already contains and/or develops a quantity of mesophase in the range 5 to 60%, for example 10 to 50%, more preferably 15 to 45% for example 20 to 45% based on the weight of the binder. The determination  
5 of the amount of mesophase present can be conducted in accordance with ASTM D 4616 – 95 (Reapproved 2000).

A particularly preferred binder comprises a material supplied by Rütgers VFT AG, Germany, under the trade mark RAUXOLIT. The RAUXOLIT is available in a variety of forms such as liquid, granules and powders. More preferably, the binder  
10 comprises RAUXOLIT FF 100 (powder). RAUXOLIT is believed to be a pitch/tar type product prepared from synthetic materials rather than from coal tar or petroleum pitch. Other materials with similar softening points to that of the RAUXOLIT FF are known, such as coal tar pitch 140 from Koppers in the US, but the latter, as the name suggests, is derived from coal. It is preferred to use a solid synthetic particulate pitch as the  
15 binder.

Aromatic polymers which can be prepared as dispersions in water, such as alkali phenolic resins may also be used as the binder. Other suitable polymers are novolak phenolic resins which are generally employed as powder mixtures with hexamine. An example of a suitable resin is the commercially available phenolic resin known as  
20 "Rutaphen F type" supplied by Bikelite, Germany.

It is preferred to use a binder containing from 0 to 50 weight %, preferably 0 to 20 wt % (based on total binder) of mesophase in the process of the present invention for making the filter material. Thus, in the present invention the bonding of the refractory particles is preferably achieved with the carbon matrix in the form of semicoke. The  
25 semicoke is preferably formed by heating coal tar or pitches, petroleum tar or pitches or synthetic aromatic polymer to cause the formation of at least some so-called "mesophase". The liquid or semi-liquid mesophase coats the surface of the refractory particles, and the mesophase is then converted on firing to form the carbon matrix of semicoke.

30 According to a preferred aspect, the invention provides a process of producing a refractory article, especially a filter for molten metal, comprising:

- (a) forming a slurry of a particulate refractory material, and a particulate binder comprising mesophase, in a liquid carrier;

- (b) forming the slurry into a desired shape; and
- (c) firing the shaped slurry.

In the manufacture of filters in accordance with the process of the present invention, thermoplasticity of the defined carbon-rich materials can produce the result that, during the firing of, for example, a moulded article comprising a mixture of a pitch, e.g. a mesophase-containing pitch, and refractory particles, the article can have a tendency to lose its shape, e.g. it can slump or distort, due to the softening of the material (which may or may not include mesophase at this stage). Any mesophase present or generated in the heated carbon-rich material also can also exhibit thermoplastic properties. Accordingly, in the present invention it is preferred to reduce the thermoplastic properties of the material, including any mesophase which may be present therein, by treating it to one or more stabilisation steps before using it as a binder in the process of the present invention.

Stabilisation of the carbon-rich materials, including any mesophase present in or generated in the pitch or the tar, or in the pyrolysed aromatic polymer, can be achieved, for example, by heat treatment thereof in the presence of air or oxygen and/or treatment with an oxidising agent such as, for example, nitric acid. The stabilisation tends to result in the carbon-rich material, including any mesophase therein, behaving more like a thermosetting resin rather than a thermoplastic material.

A preferred method of stabilising the carbon-rich material, for example a pitch and/or coal tar and/or pyrolysed synthetic phenolic resin, is heat treatment in the presence of oxygen. The carbon-rich starting material is heated (above room temperature) preferably to a temperature less than 600°C, more preferably to a temperature in the range 100-500°C, even more preferably in the range 150-400°C, e.g. approximately 180°C. The heating is conducted in the presence of oxygen, e.g. air is suitable.

A preferred method of stabilising the carbon-rich material, for example, a pitch and/or coal tar and/or pyrolysed synthetic phenolic resin is by treatment with an acid or an oxidising agent or with both. Preferably an oxidising acid is used, for example, nitric acid.

Thus in one particular embodiment of the present invention there is provided a method of producing a starting material, the method comprising oxidising a resin such as a synthetic resin, pitch or tar with an oxidising agent and heat treating the oxidised

resin thereby polymerising the resin and producing the starting material. In this particular embodiment the oxidising agent may comprise one or more acids, for example one or a combination of inorganic and/or organic acids. Alternatively the oxidising agent may comprise a non-acid oxidising agent. The oxidising agents or acids are, for example, nitric acid, hydrochloric acid, phosphoric acid, nitrous acid, chromic acid, hypochloric acid, or hydrogen peroxide. Preferably the oxidised resin is heated to a temperature up to 350°C for several hours, preferably in the absence of air. The starting material can be used as the carbon-rich stabilised material for making the filter in the process of the present invention, with or without further stabilisation.

10        Examples of suitable oxidising agents or acids include, but are not restricted to, nitric acid, hypochlorous acid, sulphuric acid, chromic acid or mixtures thereof. The treatment with oxidising agent and/or acid is preferably carried out in aqueous solution. For example nitric acid solutions containing 10-80% by volume, more preferably 15-70% by volume, provide satisfactory results. The use of concentrated nitric acid is preferred. The nitric acid treatment can be carried out, for example, by heating the carbon-rich starting material (e.g. a pitch and/or tar and/or pyrolysed synthetic phenolic resin) in the nitric acid, for example for several hours. After heat and/or nitric acid treatment, the carbon-rich material (e.g. pitch and/or tar and/or pyrolysed synthetic phenolic resin) may require grinding and/or crushing (e.g. using a ball mill) to reduce the particle size, (for example to below 50 microns in diameter) since some agglomeration may occur during such treatment.

20        As mentioned above, the binder preferably comprises a treated or untreated carbon-rich material (eg a pitch and/or tar and/or synthetic phenolic resin). It is preferred to use such materials already containing at least some mesophase, or to generate mesophase in the said material in preparing the binder. Some carbon-rich materials (e.g. pitches and/or tars and/or synthetic phenolic resin) need to be heat treated if it is desired that mesophase should be present in the binder and others may need to be so-treated if it is desired to increase the proportion of mesophase already present therein (e.g. in a pitch and/or tar and/or pyrolysed synthetic phenolic resin). However, some carbon-rich materials (eg pitches and/or tars) may already comprise a significant proportion of mesophase and hence require no pre-treatment to create or increase the amount of mesophase prior to use as a binder material in the process of the present invention. . Whilst it may be advantageous to use a binder already containing

mesophase, it is preferable that high mesophase containing binders are not employed in the present invention. If high mesophase binders are employed, the material is found to have a higher softening temperature and higher viscosity when melted which reduces its effectiveness in coating the surface of the refractory particles.

5           It has been observed above that a moulded article comprising a mixture of binder and refractory particles can have a tendency to lose its shape, e.g. it can slump or distort, due to the softening of the carbon-rich material (including any mesophase present in the binder or the mesophase formed on heating). Another method of reducing or eliminating this tendency is stabilise the carbon-rich material (including any  
10 mesophase present therein) in-situ during heat treatment or firing using a polyfunctional chemical compound, for example a polymer bearing a plurality of functional groups, preferably hydroxyl groups or carboxylate groups. One such compound that has been found particularly effective is polyvinyl alcohol (e.g. the polymer prepared by the hydrolysis of polyvinylacetate). Commercially available polyvinyl alcohol varies in its  
15 degree of hydrolysis and the degree of polymerisation. It has a large number of reactive alcohol ( $-C-OH$ ) groups along its chain length. A preferred form of polyvinyl alcohol is that supplied by Zschimmer & Schwarz GmbH & Co., Germany, under the trade mark OPTAPIX PAF 35. The material is supplied and can be used as a 35% (by weight) solution or diluted (e.g. to a 20% solution) prior to use. Other compounds that can be  
20 advantageously employed are carbomethoxy-substituted oligophenyls, carbomethoxy-substituted benzyl esters or a combination of these two compounds. The use of these compounds to reduce deformation of articles moulded using pitch as binder is disclosed in US Patent 4,877,761, in the name of Chmiel et al.

          The polyfunctional compound, e.g. the polyvinylalcohol, is preferably included  
25 in the slurry composition employed to make the filter material of the present invention. The amount of polyfunctional compound employed is preferably in the range 0.1 to 3.0 weight % more preferably 0.3 to 1.5 wt %, most preferably 0.5 to 1.0 wt% based on the weight of the solids in the slurry.

          The polyfunctional compound is believed to promote crosslinking of the binder  
30 material and thus cause stabilisation of the shape of the formed or moulded filter during the heating and firing thereof.

          The polyfunctional compound can be employed to assist in stabilising the binder regardless of whether the binder material has been subjected to other stabilisation

methods, e.g. nitric acid treatment or oxidation with air at elevated temperature.

The liquid carrier in the slurry can be any suitable liquid diluent, for example water, methanol, ethanol, light petroleum. However, water is generally the preferred carrier as it provides slurries having good coating properties and is environmentally  
5 safe.

Other materials that can be incorporated in the slurry comprising the binder and the refractory particles used in the present invention include suspension aids, anti-foaming agents, humectants and dispersion agents. The use of such materials in the preparation of filters using slurry coating of a disposable former is known in the art. The  
10 concentrations of such additives, if any, in the slurry, based on the total weight of solids can be, for example, as follows:

Suspension aids – 0 to 1.0 wt%, e.g. 0.1 wt%

Anti-foaming agents – e.g. silicone anti-foam – 0 to 1.0 wt%, e.g. 0.3 wt%

Polymeric stabiliser – e.g. polyvinyl alcohol solution – 0 to 10.0 wt%, e.g. 3.0  
15 wt% of a 20% aqueous solution

Humectants – 0 to 1.0 wt%, e.g. 0.5 wt%

Dispersants – e.g. ammonium lignosulphonate – 0 to 1.0 wt%, e.g. 0.6 wt%

The dispersant, ammonium lignosulphonate and the polymer material, polyvinyl alcohol also function together as so-called ‘green binders’. For the  
20 manufacture of certain refractory articles such as filters, these ‘green binders’ ensure that the refractory powders and the binder remain on the disposable former, e.g. the polyurethane foam after drying without crumbling away or cracking and that the unfired filter can be handled and can tolerate further processing stages, e.g. spraying, without damage occurring.

25 The quantity of liquid present in the slurry is preferably such that the relative proportions (in percentage by weight) of the total “solids” to liquid in the slurry are in the range at least 50% solids : no more than 50% liquid; more preferably they are in the range at least 55% solids : no more than 45% liquid; even more preferably they are in the range at least 60% solids : no more than 40% liquid. For example, the slurry may  
30 comprise approximately 69% solids and 31% liquid.

The disposable former provides as a template for the desired shape of the open-pored porous material which is produced by the process of the present invention. By “disposable” is meant that the material from which the former is fabricated is degraded

and is substantially lost to the atmosphere by combustion or volatilisation when the coated former is fired. The former can be, for example, a three dimensional lattice-work built up from reticulated layers, an extruded net compressed to form an open reticulated structure, or a reticulated polymeric foam. Preferably the disposable former is

5 reticulated polyurethane foam. Reticulated polyurethane foams suitable for use as disposable formers for molten metal filters are well known in the art. Suitable foam material is supplied by 1. Kureta GmbH & Co., D-35260 Stadtallendorf, Germany; 2. Eurofoam Deutschland GmbH, Troisdorf, Germany and 3. Caligen Europe B.V., Breda, Netherlands.

10 The coating is preferably applied to the disposable former using means well known in the art. Thus, for example, a polyurethane foam former can be dipped in or sprayed with the slurry and then passed between pairs of rollers to adjust the distribution and amount of slurry present on the foam.

Thus, in a preferred method of forming a filter in accordance with the present  
15 invention the filter may, for example, be formed in a conventional manner known to the skilled person, for example, by impregnating a polymeric (usually polyurethane) foam with a slurry (preferably an aqueous slurry) of a composition comprising a particulate refractory material and a binder, and drying and firing the impregnated foam to drive off the liquid, to harden the composition, and to burn off the polymeric foam, thereby  
20 creating a refractory foam structure which may be used as a filter, e.g. for molten metal. The step of forming the slurry into the desired shape, preferably therefore comprises impregnating a polymeric foam with the slurry (e.g. by spraying the slurry onto and into the foam and/or by rolling the slurry onto and into the foam). The final steps of the process preferably comprise drying and firing the impregnated foam to drive off the  
25 liquid and any other volatiles, to harden the refractory/binder composition, and to burn off the polymeric foam.

The preferred filter forming method according to the present invention differs from conventional methods, however, in that the composition includes the defined binder and refractory particle composition, preferably comprising mesophase in the  
30 binder, and the temperature at which the impregnated foam is fired is generally lower than conventional firing temperatures. As already mentioned, the firing temperature according to the invention is preferably no higher than 800°C, more preferably no higher than 700°C, even more preferably no higher than 650°C, for example

approximately 600°C. For example, the filter may be fired for a total of 5 hours at heating and cooling rates (from and to ambient temperature) of approximately 300°C per hour, with a maximum firing temperature of 600°C (which the filter may experience, for example, for approximately 1 hour).

- 5           After the disposable former has been coated, it is dried and then subjected to further coating if desired until the requisite thickness of slurry has been applied.

          The coated former is then preferably heated to a higher temperature to drive off any residual volatiles, and is finally fired at a temperature at least sufficient to convert the binder to a carbon matrix, preferably comprising the semi coke. It is preferred to fire  
10   the coated former in the absence of oxygen.

          The firing of the article substantially in the absence of oxygen generally causes some or all of the carbon-rich material (i.e. the binder), comprising at least some mesophase which is either already present, or is generated by the heating, to pyrolyse (at least partially), generally forming the material known as "semi coke". This semi coke  
15   forms a matrix which bonds the particulate refractory particles together, forming an article comprising refractory material and carbon as the major components. In the fired refractory article, some or all of the binder is derived from mesophase, and will generally comprise semi coke. Articles containing mesophase or those that generate mesophase are preferably fired in the absence of oxygen.

- 20           The articles made by the processes of the present invention are produced by firing the shaped composition, with or without a disposable former, comprising refractory particles and binder at a temperature preferably no higher than 800°C, more preferably no higher than 700°C, even more preferably no higher than 650°C, for example approximately 600°C. The firing to form the article is preferably carried out  
25   substantially in the absence of oxygen, e.g. in an inert or "non-oxidising" atmosphere, such as nitrogen or argon for example, or under vacuum, or under a "reducing atmosphere", such as hydrogen and/or carbon monoxide and/or coal gas (i.e. a mixture of methane and hydrogen). Additionally or alternatively, the firing of the shaped slurry may be carried out in the presence of a reducing agent, for example carbon (graphite), to  
30   scavenge some or all of the oxygen which may be present. The shaped slurry or coated former is normally at least partially dried, e.g. at a temperature between 100°C and 200°C (e.g. approximately 150°C) prior to being fired, but it may be fired without significant pre-drying.

The firing of the article will normally be carried out in an oven or furnace, but additionally or alternatively other forms of heating may be used, for example microwave or radio frequency heating.

5 In a further embodiment of the present invention the firing of the shaped filter material is carried out in two stages. It has been found that this two stage process leads to the production of filter materials which have improved compressive strength.

In this embodiment of the present invention the unfired filters comprising the shaped slurry are subjected to a two-step firing protocol:

- 10 1. the filters are placed in a container through which a controlled supply of air is passed and heated gradually to a temperature in the range 340 to 360°C, for example at a heating rate of 60 to 100°C per hour
2. the air supply is turned off and the heating continued at a rate of approximately 200°C per hour up to a temperature in the range 675 to 725°C then holding at this temperature for a further period of 50 to 70  
15 minutes.

In this two-step firing process the binder is preferably a synthetic pitch, most preferably the commercially available RAUXOLIT material referred to above.

Thus, for example, it has been found that good results are obtained when conducting the firing as follows. A number of 'green', unfired filters, comprising  
20 reticulated polyurethane foam impregnated with a slurry containing the refractory particles and the mesophase-containing, or mesophase generating binder, are placed on open mesh metal trays and then placed on a metal rack which holds a number of trays. The rack stands on a shallow metal trough and a metal cover is lowered down over the rack which makes a tight seal with the trough at the bottom. Graphite is also sprinkled  
25 into the trough to react with any air which may leak between the cover and the trough. This then constitutes the firing box for the filters. In the initial part of the firing cycle from room temperature to 350°C, at a rate of temperature increase between 60 – 100°C per hour, air is blown into the box through at a rate of 15 litres per minute. When the temperature inside the box reaches 340 – 350°C, the air supply is turned off.  
30 Measurements made with an oxygen meter show that during the time that air is blown into the box the oxygen level inside is between 2 and 6%. When the air is stopped the oxygen level immediately drops to zero.

It is believed that the air in the box during the first part of the firing reacts with

the binder in such a manner that a higher yield of carbon is produced when the final firing takes place.

In a firing conducted without injecting air, the filter compression strengths, measured using a Hounesfield Tensometer, averaged only 300 – 500 newtons, whereas  
5 those fired with air injected, had compression strengths averaging 700 – 800 newtons. During firing the filters lose weight due to the loss of volatile organic compounds. Without air the loss is 22%, whilst with air, it is 18% indicating a higher coke yield during firing and consequently higher strength.

The filters of the present invention can be made advantageously with lower  
10 density and lower thermal mass than conventional prior art filters for steel (for example formed from glass-bonded zirconia), and because of this, the filter abstracts less heat from the metal during pouring. Consequently, steels such as plain carbon steel which is often poured at temperatures not significantly higher than the liquidus temperature generally do not have to be superheated in order for the metal to pass through the filter  
15 without freezing. Therefore, for example, plain carbon steel may be poured at much lower temperatures than conventionally. This provides the economic and environmental benefits that less energy needs to be used.

The process for making the filter material according to the present invention has the advantages of requiring less energy consumption, being easy to use, requiring  
20 relatively low pressures and low temperatures, and being capable of producing very thin sections if required (for example in the order of 0.3 mm). Moreover, the preferred process is water based and the processing time is relatively short.

It is possible to make other refractory articles using the compositions disclosed above for forming the porous filter. Thus, compositions comprising the refractory  
25 particles and the defined binder, can be used to make a variety of other refractory products. An example of such a product is a feeder sleeve, for retaining a reservoir of molten metal for supplying to a mould cavity as a metal casting within the cavity shrinks as it cools and solidifies.

For the avoidance of doubt, it is to be understood that features (including  
30 material compositions) described with reference to a particular aspect of the invention are applicable to each aspect of the invention.

The invention will now be further described with reference to the following Examples.

### Example 1

Five refractory filler systems were prepared using the RAUXOLIT FF 100 commercially available binder as follows:

- 1.a Rauxolit: graphite
- 5      1.b Rauxolit: alumina
- 1.c Rauxolit: alumina/graphite ( $\text{Al}_2\text{O}_3$ :graphite = 3:1)
- 1.d Rauxolit: silicon carbide
- 1.e Rauxolit: zircon.

A number of 100 x 100 x 25 mm filters were produced from these formulations.

- 10      The percentage of refractory particles ranged from 55 to 75% and the RAUXOLIT from 45 to 25%. The firing temperatures ranged from 400 to 600°C.

- 15      These filters were then tested with molten steel using a direct impingement test in which 100 kg of high alloy steel, typically Cr8M grade, at a temperature of 1600 – 1610°C is poured from a bottom-pour ladle, fitted with a 29 mm nozzle, down a 700 mm sprue onto the face of a filter, which is supported on two opposing sides in a print in a resin-bonded sand mould. The test provides a measure of the mechanical strength of the filter from the initial metal impact, thermal shock resistance, the mechanical strength at temperature, the resistance to chemical attack from the chemically aggressive alloy and the resistance to erosion from the very fluid steel.

- 20      Two recipes performed the best in this test, the alumina/graphite and zircon recipes with RAUXOLIT levels of 35 and 25% respectively fired at 600°C. Neither filter recipes exhibited any cracks or erosion.

Of the two recipes, however, the easiest to process was the alumina/graphite recipe.

### 25      Example 2

- For the manufacture of a carbon bonded graphite filter for molten aluminium filtration, approximately 40% by weight of nitric acid-treated RAUXOLIT FF 100 binder was employed. The nitric acid treatment comprised adding 20wt% of concentrated nitric acid to the RAUXOLIT FF 100 powder and mixing thoroughly for 1  
30      to 2 hours. The mixture was then heated in an oven to a temperature in the range 300 to 350°C for approximately 12 to 18 hours and then allowed to cool. The product was then washed with water to remove any nitric acid residues and then dried in an oven at 120°C. The resultant lumpy product was then crushed in a ball mill.

The nitric acid and heat-treated RAUXOLIT FF 100 binder was mixed with approximately 60wt% graphite powder. The slurry was made by adding approximately 40wt% water (based on weight of solid binder and refractory) and up to a total 2 wt% (based on weight of solid binder and refractory material) of organic binder, thickener  
5 and suspension agent (polyvinyl alcohol solution, xanthan gum and ammonium lignosulphonate).

This slurry was used to impregnate a polyurethane foam slab of dimensions 50 mm x 50 mm x 15mm. On firing to 600°C for 1 hour at a firing rate of 300°C per hour in a metal box to exclude air, a product with an open, porous structure was obtained  
10 suitable for use as a filter. Such a filter was used successfully in filtering aluminium at a temperature of 800°C, and gave a comparable performance to a conventional glass bonded commercial ceramic filter (supplied by Foseco under the trade name SIVEX FC) manufactured by firing at a temperature in the range 1000 to 1100°C.

### Example 3

15 For the manufacture of a carbon bonded alumina and graphite filter for iron filtration, approximately 30% by weight of nitric acid and heat-treated RAUXOLIT FF 100 binder, prepared as in Example 2, was mixed with approximately 50% alumina powder and approximately 20% graphite powder. Slurry was made by adding approximately 40 wt% water (based on weight of solid binder and refractory) and up to  
20 2wt% (based on weight of solid binder and refractory material) organic binder, thickener and suspension agent (polyvinyl alcohol solution, xanthan gum and ammonium lignosulphonate). This slurry was used to impregnate a polyurethane foam slab of dimensions 75 mm x 75 mm x 20 mm. Upon firing to 600°C for one hour at a heating rate of 300°C per hour in a metal box to exclude air, a porous material suitable  
25 for use as a metal filter was obtained. This filter was used successfully to filter molten iron at 1,450°C and gave a comparable performance to a conventional glass bonded commercial ceramic filter (supplied by Foseco under the trade name SEDEX) manufactured by firing at a temperature in the range 1100 to 1200°C.

30

Claims

1. A filter suitable for filtering molten metal comprising an open-pored porous material comprising particles of refractory material embedded in and bonded together by a bonding material comprising a carbon matrix.
2. A filter as claimed in Claim 1 wherein the refractory material is selected from  
5 zirconia, zircon, silica, alumina, titania, silicon carbide, zirconium carbide, titanium carbide, calcium carbide, aluminium carbide, silicon nitride, aluminium nitride, nickel oxide, chromic oxide, magnesia, mullite, graphite, anthracite, coke, active carbon, graphite-magnesia, graphite-alumina, graphite-zirconia, and mixtures comprising two or more of these.
- 10 3. A filter as claimed in Claim 1 or 2 wherein the relative proportion of refractory to bonding material by weight is at least 50% refractory: no more than 50% bonding material.
4. A filter as claimed in any one of Claims 1 to 3 wherein the relative proportion by weight of refractory to bonding material lies in the range 65 to 75% refractory: 35 to  
15 25% bonding material.
5. A filter as claimed in any one of the preceding Claims wherein the pores in the open-pored porous material comprise a series of parallel ducts passing through the material.
6. A filter as claimed in any one of Claims 1 to 4 wherein the pores in the open-  
20 pored porous material comprise a random distribution of irregular interconnecting passages.
7. A filter as claimed in any one of the preceding Claims wherein the bonding material is a coke or semicoke matrix.

8. A process for making a material suitable for filtering molten metal comprising forming into an open-pored porous material a mixture comprising a binder and refractory particles and firing the material, wherein the binder is a carbon-rich source selected from one or more of the following: pitch, tar, and aromatic organic polymer that degrades to form carbon on pyrolysis.
9. A process as claimed in Claim 8 wherein the open-pored porous material is formed by compressing a mixture of the binder and the refractory particles into a disk or slab in a die and piercing the compressed material with a plurality of needles or rods to create perforations traversing the thickness of the disc or slab.
10. A process as claimed in Claim 8 wherein the open-pored porous material is formed by extruding a mixture comprising the binder and the refractory particles through a suitable die equipped with a plurality of mandrels to form perforations in the extruded article.
11. A process of producing a refractory article, especially a filter for molten metal, comprising
- (a) forming a slurry of a particulate refractory material, and a particulate binder comprising mesophase, in a liquid carrier;
  - (b) forming the slurry into a desired shape; and
  - (c) firing the shaped slurry.
12. A process for making an open-pored porous material suitable for filtering molten metal comprising
- (1.) forming a slurry comprising (a) particles of a refractory material, (b) a binder and (c) a liquid carrier,
  - (2.) coating a disposable former with the slurry
  - (3.) drying the coated former,
  - (4.) optionally applying one or more additional coats of a refractory material and/or a binder, optionally with liquid carrier, and drying the one or more additional coats,
  - (5.) firing the coated former to produce a porous material,
- wherein the binder is a carbon-rich source selected from one or more of the following classes of materials: pitches, tars, and organic polymers that degrade to form carbon on pyrolysis.
13. A process for making an open-pored porous material suitable for filtering

14. molten metal comprising
- (1.) forming a slurry comprising (a) particles of a refractory material, (b) a binder and (c) a liquid carrier,
  - (2.) coating a disposable former with the slurry
  - 5 (3.) drying the coated former,
  - (4.) optionally applying one or more additional coats of a refractory material and or a binder, optionally with liquid carrier, and drying the one or more additional coats,
  - (5.) firing the coated former to produce a porous material,
- 10 wherein the binder is a carbon-rich source selected from one or more of the following classes of materials: pitches, tars, and organic polymers that degrade to form carbon on pyrolysis, and wherein (i) the binder has been subjected to stabilisation by pre-treatment with an acid and/or with an oxidising agent and/or (ii) a polyfunctional organic compound is included in the slurry or the coating to promote stabilisation of the
- 15 binder.
14. A process as claimed in any one of Claims 8 to 13 wherein the refractory is selected from zirconia, zircon, silica, alumina, titania, silicon carbide, zirconium carbide, titanium carbide, calcium carbide, aluminium carbide, silicon nitride, aluminium nitride, nickel oxide, chromic oxide, magnesia, mullite, graphite, anthracite,
- 20 coke, active carbon, graphite-magnesia, graphite-alumina, graphite-zirconia, and mixtures comprising two or more of these.
15. A process as claimed in any one of Claims 8 to 13 wherein the refractory is provided by a mixture of graphite and one or more other particulate refractory materials, and wherein the graphite content is 10 to 40 weight % based on the total weight of the
- 25 refractory particles.
16. A process as claimed in any one of Claims 8 to 15 wherein the binder is a carbon-rich source selected from coal tar, petroleum pitch, asphalt, bitumen, synthetic pitch, synthetic tar, synthetic bitumen; or residues derived from the pyrolysis of coal, crude oil, coal tar, petroleum pitch, asphalt, bitumen, synthetic pitch, synthetic tar or
- 30 synthetic bitumen.
17. A process as claimed in any one of Claims 8 to 16 wherein the binder is a solid synthetic particulate pitch.
18. A process as claimed in any one of Claims 8 to 17 wherein the carbon-rich

source is an aromatic organic polymer.

19. A process as claimed in any one of Claims 8 to 18 wherein the binder is subjected to stabilisation by heat treatment with nitric acid.

20. A process as claimed in Claim 13 wherein the polyfunctional compound is  
5 polyvinylalcohol.

21. A process as claimed in Claim 12 or 13 wherein the disposable former is reticulated polyurethane foam.

22. A process as claimed in Claim 11, 12 or 13 wherein the liquid carrier is water.

10

15

20

25

30

## INTERNATIONAL SEARCH REPORT

International Application No  
PCT/GB 01/03846

## A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 B22C9/08 B22D43/00 C04B38/00 C04B38/06

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 B22C B22D C04B

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, WPI Data

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	EP 0 308 928 A (PETOCA LTD) 29 March 1989 (1989-03-29) page 2, line 9 - line 12 page 2, line 54 - line 57; claim 2 examples 1,2	1-3,6-8
Y	---	12-22
Y	US 4 265 659 A (BLOME JAMES C) 5 May 1981 (1981-05-05) abstract column 2, line 1 - line 5 column 2, line 16 - line 19 column 2, line 35 - line 47 column 2, line 58 - line 61; example 1	12-22
Y	US 3 893 917 A (PRYOR MICHAEL J ET AL) 8 July 1975 (1975-07-08) example 1 ---	12-22
	--- -/--	

☒ Further documents are listed in the continuation of box C.

☒ Patent family members are listed in annex.

## \* Special categories of cited documents:

- \*A\* document defining the general state of the art which is not considered to be of particular relevance
- \*E\* earlier document but published on or after the international filing date
- \*L\* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- \*O\* document referring to an oral disclosure, use, exhibition or other means
- \*P\* document published prior to the international filing date but later than the priority date claimed

- \*T\* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
- \*X\* document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
- \*Y\* document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.
- \*&\* document member of the same patent family

Date of the actual completion of the international search

17 December 2001

Date of mailing of the international search report

18/01/2002

Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2  
NL - 2280 HV Rijswijk  
Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,  
Fax: (+31-70) 340-3016

Authorized officer

Wirth, S

# INTERNATIONAL SEARCH REPORT

In International Application No

PCT/GB 01/03846

## C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	EP 0 708 064 A (DOLOMITWERKE GMBH) 24 April 1996 (1996-04-24) cited in the application abstract -----	11
A	WO 99 65843 A (ALLIED SIGNAL INC) 23 December 1999 (1999-12-23) the whole document -----	1-22

# INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No  
PCT/GB 01/03846

Patent document cited in search report		Publication date	Patent family member(s)	Publication date
EP 0308928	A	29-03-1989	JP 1079073 A	24-03-1989
			DE 3865616 D1	21-11-1991
			EP 0308928 A1	29-03-1989
			US 5091164 A	25-02-1992
US 4265659	A	05-05-1981	AT 2874 T	15-04-1983
			CA 1153019 A1	30-08-1983
			DE 3000835 A1	30-04-1981
			DE 3062543 D1	05-05-1983
			EP 0029802 A1	03-06-1981
			JP 1759041 C	20-05-1993
			JP 4031728 B	27-05-1992
			JP 56060619 A	25-05-1981
			NO 802972 A ,B,	10-04-1981
			US 4342664 A	03-08-1982
US 3893917	A	08-07-1975	US 4056586 A	01-11-1977
			US 3947363 A	30-03-1976
EP 0708064	A	24-04-1996	DE 4437308 A1	25-04-1996
			EP 0708064 A2	24-04-1996
			JP 2802343 B2	24-09-1998
			JP 8210781 A	20-08-1996
WO 9965843	A	23-12-1999	US 6126874 A	03-10-2000
			AU 6128999 A	05-01-2000
			CN 1310697 T	29-08-2001
			EP 1084089 A2	21-03-2001
			WO 9965843 A2	23-12-1999
			US 6146611 A	14-11-2000